

9-(3,4-Dimethoxyphenyl)-4-*p*-tolyl-4,5,6,7-tetrahydrofuro[3,4-*b*]quinoline-1,8(3*H*,9*H*)-dione

Shu-Jiang Tu,* Yan Zhang and Run-Hong Jia

Department of Chemistry, Xuzhou Normal University, Xuzhou 221116, People's Republic of China

Correspondence e-mail: laotu2001@263.net

The title compound, C₂₆H₂₅NO₅, was synthesized by the reaction of 3,4-dimethoxybenzaldehyde, 3-(*p*-tolylamino)-cyclohex-2-enone and tetronic acid in glacial acetic acid under microwave irradiation. Its molecular structure shows a planar furan ring, a dihydropyridine ring in a flattened envelope conformation and a cyclohexenone ring in an envelope conformation.

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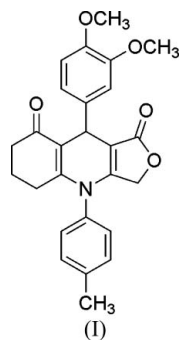
Key indicators

Single-crystal X-ray study
T = 298 K
 Mean σ (C–C) = 0.005 Å
R factor = 0.060
wR factor = 0.153
 Data-to-parameter ratio = 12.9

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

Tetronic acid derivatives and their metabolites are widespread in nature, of which vitamin C and penicillic acid are undoubtedly the most important (Neelakantan & Seshadri, 1959). Natural 4-ylidenetetronic acid derivatives known as pulvinic acids have been found as pigments in lichens and higher fungi (Weinstock *et al.*, 1979). Tetronic acid derivatives are interesting because of their antibiotic, antitumor, anti-coagulant, anti-epileptic, antifungal, and anti-inflammatory properties (Foden & McCormick, 1975). We report here the crystal structure of the title compound, (I).



The dihydropyridine ring adopts a flattened envelope conformation, with atom C5 deviating from the C1/C4/C6/C11/N1 plane by 0.174 (3) Å. The cyclohexenone ring adopts an envelope conformation (Fig. 1), with atom C9 deviating from the C6/C7/C8/C10/C11 plane by 0.648 (6) Å. The dihedral angles between the C1/C4/C6/C11/N1 plane and the C12–C24 and C19–C24 benzene ring planes are 74.1 (2) and 85.0 (1)°, respectively. The cyclohexenone and the furanone rings make dihedral angles of 6.4 (2) and 4.7 (2)°, respectively, with the central C1/C4/C6/C11/N1 plane.

Experimental

Compound (I) was prepared by the reaction of 3,4-dimethoxybenzaldehyde (1 mmol), 3-(*p*-tolylamino)cyclohex-2-enone (1 mmol) and tetronic acid (1 mmol) in glacial acetic acid (2 ml) under microwave irradiation. Single crystals of (I) suitable for X-ray

diffraction were obtained by slow evaporation of a 95% aqueous ethanol solution (yield 93%; m.p. 499 K).

Crystal data

$C_{26}H_{25}NO_5$	$V = 1085 (2) \text{ \AA}^3$
$M_r = 431.47$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.320 \text{ Mg m}^{-3}$
$a = 8.427 (9) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 11.453 (12) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$c = 11.579 (12) \text{ \AA}$	$T = 298 (2) \text{ K}$
$\alpha = 95.909 (14)^\circ$	Block, yellow
$\beta = 100.870 (13)^\circ$	$0.22 \times 0.18 \times 0.09 \text{ mm}$
$\gamma = 94.902 (15)^\circ$	

Data collection

Bruker SMART CCD 1000 area-detector diffractometer	5706 measured reflections
φ and ω scans	3767 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	1874 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.980, T_{\max} = 0.992$	$R_{\text{int}} = 0.031$
	$\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.060$	$w = 1/[\sigma^2(F_o^2) + (0.0634P)^2]$
$wR(F^2) = 0.153$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.99$	$(\Delta/\sigma)_{\text{max}} < 0.001$
3767 reflections	$\Delta\rho_{\text{max}} = 0.20 \text{ e \AA}^{-3}$
292 parameters	$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$

H atoms were positioned geometrically and refined as riding, with C—H = 0.93–0.98 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{methyl C})$ or $1.2U_{\text{eq}}(\text{C,N})$.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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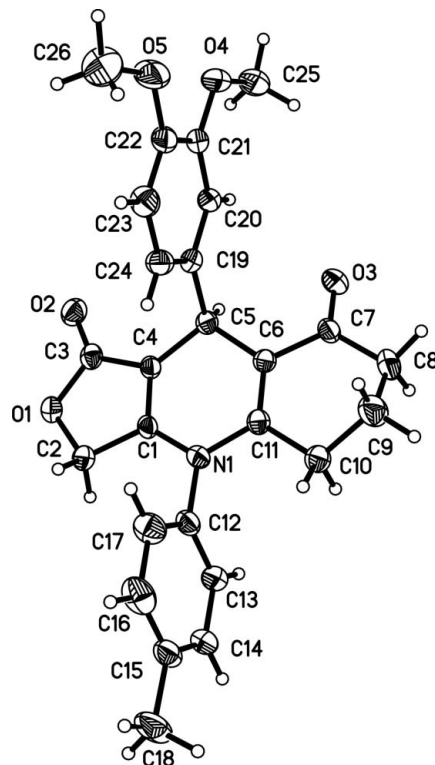


Figure 1 The structure of (I), showing 30% probability displacement ellipsoids.

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